Water Supply Zone: St Albans North (AF088) Period: 01-Jan-2023 to 31-Dec-2023

Population: 54184



Davamatav	Haita	No. of	PCV	No. of Samples >PCV	% of Samples >PCV	N#:	Mean	Max.
Parameter	Units	Samples	icrobiological Parameters	>PCV	>PCV	Min.	Mean	iviax.
Coliform bacteria	No./100ml	132	0	0	0	0	0	0
E coli	No./100ml	132	0	0	0	0	0	0
Clostridium perfringens	No./100ml	8	0	0	0	0	0	0
Enterococci	No./100ml	8	0	0	0	0	0	0
3 day plate count 22 °C	No./1ml at 22 °C	52	No abnormal change	0	0	0	3	132
3 day plate count 22 C	NO./1111 at 22 C	32	Customer Parameters	0	-	- 0		132
Alkalinity	mgHCO3/I	1	No PCV	0	0	338	338	338
Calcium	mgCa/I	1	No PCV	0	0	133	133	133
Chlorine (Residual)	mgCl2/l	132	No PCV	0	0	0.1	0.35	0.74
Colour	mg/I Pt/Co	52	20	0	0	<2.5	<2.5	<2.5
Fluoride	mgF/I	8	1.5	0	0	0.072	0.084	0.094
Hardness (Total)	mgCaCO3/I	1	No PCV	0	0	333	333	333
	- '	52	6.5-9.5	0	0	6.9	333 7	7.2
Hydrogen Ion (pH) Quantitative Odour	pH value Dilution No.	52	Abnormal & unacceptable to	0	0	0.9	0	0
		52	·	0	0	0	0	0
Quantitative Taste	Dilution No.	132	consumers No PCV	0	0	7.4	13.8	21.7
Temperature	°C			0	0			
Turbidity	NTU	52	4	U	0	<0.10	<0.10	0.28
NA-4-1-			Chemicals					
Metals	A /I	8	10	0	0	<1.0	<1.0	-1.0
Arsenic	μgAs/I	8 8	200	0	0			<1.0
Aluminium	μgAI/I					<5.0	<5.0	<5.0
Antimony	μgSb/l	8	5	0	0	<0.20	<0.20	<0.20
Cadmium	μgCd/l	8	5	0	0	<0.20	<0.20	<0.20
Chromium	μgCr/l	8	50	0	0	<0.5	0.6	2.2
Copper	mgCu/l	8	2	0	0	<0.019	0.061	0.224
Iron	μgFe/l	8	200	0	0	<15.0	<15.0	<15.0
Lead	μgPb/l	8	10	0	0	<1.00	1.27	4.58
Manganese	μgMn/l	8	50	0	0	<1.0	<1.0	<1.0
Mercury	μgHg/l	8	1	0	0	<0.10	<0.10	<0.10
Nickel	μgNi/l	8	20	0	0	<2.0	<2.0	<2.0
Sodium	mgNa/l	8	200	0	0	13.4	16.1	17.9
Pesticides								
Atrazine	μg/l	8	0.1	0	0	<0.013	0.02	0.027
Total Pesticide	μg/l	8	0.5	0	0	0.013	0.041	0.052
Additional Parameters								
Ammonium	mgNH4/l	8	0.5	0	0	<0.05	<0.05	< 0.05
Benzene	μg/l	8	1	0	0	<0.07	<0.07	<0.07
Benzo (a) Pyrene	μg/l	8	0.01	0	0	< 0.001	<0.001	<0.001
Boron	mgB/I	8	1	0	0	< 0.100	<0.100	<0.100
Bromate	μgBrO3/l	8	10	0	0	<0.3	<1.5	<1.5
Chloride	mgCl/l	8	250	0	0	32	34	36
Electrical Conductivity at 20 °C	μS/cm at 20 °C	52	2500	0	0	578	618	724
Nitrate	mgNO3/I	8	50	0	0	22.9	29.4	34.3
Nitrite	mgNO2/l	8	0.5	0	0	< 0.007	< 0.007	< 0.007
Nitrite Nitrate Formula	=	8	1	0	0	< 0.46	< 0.69	< 0.69
Selenium	μgSe/l	8	10	0	0	<1.0	<1.0	<1.0
Sulphate	mgSO4/I	8	250	0	0	19	24	30
Sum of Tri & Tetrachloroethene	μg/l	8	10	0	0	0	0	0
Tetrachloromethane	μg/l	8	3	0	0	<0.2	<0.2	<0.2
Total Cyanide	μgCN/I	8	50	0	0	<1.2	<1.2	<1.2
Total Organic Carbon	mgC/I	8	No abnormal change	0	0	0.4	0.5	0.6
Total PAHs	μg/l	8	0.1	0	0	0.4	0.5	0.0
Total Trihalomethanes	μg/l	8	100	0	0	1.28	5.29	12.21
rotal iriilaloilletilalles	ug/I	O	100	U	U	1.20	3.23	14.41

## Notes

PCV = Prescribed Concentration or Value or Specification Concentration or Value

## **Commentary on Water Quality**

Water quality was satisfactory in this zone in 2023.

## **Undertakings & Authorised Departures**

No Undertakings or Authorised Departures applied to this water supply zone during 2023.

The DWI has served the Company with a Notice relating to PFOS in the Holywell WTW supply system which includes this water supply zone. The Company has committed to replacing the granular activated carbon (GAC) in the contactors at Holywell WTW to enhance PFOS removal. This work is scheduled to be completed by March 2026.

## Glossary Drinking Water Standards



The report above show all regulatory parameters which are monitored in accordance with the current Water Supply (Water Quality) Regulations. Some no....

Regulatory Parameters		
Parameter	What it means	Standard
Microbiological Parameters		
	These bacteria are widely distributed in the environment and provide a sensitive measure of the	
	microbiological quality of the water supply. They are removed during the treatment process. However, if any	
	coliform organisms are detected in drinking water immediate action is taken to investigate the source of the	
	bacteria. Nearly all instances of coliforms in samples taken from customers' taps are due to microbiological	
Coliform bacteria	growths in the domestic cold taps.	0 per 100ml
E coli		
Clostridium perfringens		
Enterococci	Bacteria which are indicative of possible faecal contamination. Immediate action is taken if these organisms are	
Enterococci	detected in drinking water.	0 ner 100ml
	detected in drinking water.	0 per 100ml
		standard
2 day plate count 37 °C	A range of harmless bacteria that may be present in water supplies. These are monitored to ensure the	(increasing trends
3 day plate count 22 °C	efficiency of the treatment process and the cleanliness of the distribution system.	are investigated)
Customer Parameters	emelency of the treatment process and the cleaniness of the distribution system.	are investigated)
Customer Farameters		
	All clinits is normally due to bissylven stocalty of calcium and magnetium but you apparently so diver-	
	Alkalinity is normally due to bicarbonate salts of calcium and magnesium, but very occasionally sodium	
A Harding San	bicarbonate may contribute. In the former case the alkalinity is sometimes called the "temporary hardness" as	No specific
Alkalinity	it is removed by boiling.	standard
	Occurs naturally in water after passage through mineral deposits and rock strata. Calcium contributes to the	No specific
Calcium	total hardness of water.	standard
	Affinity Water disinfects some of our water supplies using chlorine. The concentration of chlorine used is	
	carefully controlled and is set to ensure that water is adequately disinfected, while minimising any taste or	No specific
Chlorine (Residual)	odour issues for consumers.	standard
ESime (Mesidadi)	Sasar issues for consumers.	Standalu
	Water should be clear and bright, but natural organic matter or pipework corrosion products may occasionally	
Colour	impart a slight tint. The standard is set for reasons of appearance and requires water to be virtually colourless.	20 mg/l Pt/Co
	Occurs naturally in many water sources. The standard is set to ensure no adverse effects. Affinity Water does	
Fluoride	not artificially fluoridate the water supplies.	1.5 mg F/l
	Hardness is due to calcium and magnesium salts dissolved in the water. Hard water is perfectly safe and there is	
	evidence that it can even be good for your health, unless there are specific requirements to do so there is no	
		N:6:-
	need to soften the water. Almost all Affinity Water supplies are hard due to the natural geology of Southern	No specific
Hardness (Total)	England.	standard
	A measure of the acidity or alkalinity of water; pH <7.0 is acidic and pH >7.0 is alkaline. Excessively acidic or	Min. 6.5 to max
Hydrogen Ion (pH)	alkaline water can contribute to corrosion of pipes and fittings.	9.5
		Abnormal &
Quantitative Odour	Specialist tasting panels examine the water for taste or odour. These standards are measure of the aesthetic	unacceptable to
Quantitative Taste	quality of drinking water. Unusual odours or tastes may indicate a problem which needs investigating.	consumers
Temperature		No specific
	The standard requires that there should be no haziness caused by fine particles. Sometimes minute air bubbles	
	give the supply a milky appearance but on standing for a few minutes these will clear from the bottom of the	4 NTU
Turbidity	glass upwards.	
Chemicals	!* · ·	
Metals		
		F Cl #
Antimony		5 μg Sb/l
Arsenic		10 μg As/l
Cadmium	Very low levels of these substances may occur naturally, but in higher amounts could be associated with	5 μg Cd/l
Chromium	industrial pollution. The standards are health-related and have a large safety factor built in.	50 μg Cr/l
Mercury	<del></del>	1 μg Hg/l
Nickel	<del></del>	20 μg Ni/l
		20 μg INI/I
	Converse potential in management of the converse of the conver	
	Occurs naturally in many water resources. Aluminum compounds are also used at some water treatment	
Aluminum	works to remove impurities, but are themselves removed in the process.	200 μg Al/l
	Any significant amount of copper is likely to come from corrosion of customers' pipes and fittings. An excess of	
Copper	copper can cause a metallic taste.	2 mg Cu/l
	· · ·	3 ,
	Iron may be associated with corrosion of old iron water mains. Iron based compounds are also used at some	
	· ·	
Iron	water treatment works to remove impurities, but are themselves removed in the process. The standard for iron has been set for aesthetic reasons as levels persistently above the standard can give rise to discoloured water.	200 μg Fe/l

	Absent in the water entering supply but variable concentrations of lead may be found in water at the	
	customer's tap in older properties built at a time when lead was commonly used in domestic plumbing	
Lead	systems. The standard recognises that the intake of lead should be minimised for public health reasons.	10 μg Pb/l
	Occurs naturally in many waters but is usually removed during treatment. The standard is set for aesthetic	_
Manganese	reasons as black deposits of manganese dioxide can cause discoloured water.	50μg Mn/l
	May be naturally present after passing through certain mineral deposits and rock strata or introduced by some	
Sodium	water softening processes. The standard is set well below the level which could affect health.	200 mg Na/l
Pesticides		, , , , , , , , , , , , , , , , , , ,
Atrazine		0.1 μg/l
Carbetamide		0.1 μg/l
Clopyralid		0.1 μg/l
Glyphosate		0.1 μg/l
Mecoprop Metaldehyde	Associated with the use of these substances by agriculture, industry and local authorities. The standards are set well below the levels that might cause health problems, but levels should be minimised by good practice and	0.1 μg/l
Metazachlor	appropriate controls. We measure the wide range of substances that may be present.	0.1 μg/l 0.1 μg/l
Propyzamide	appropriate controls. We incasure the wide range of substances that may be present.	0.1 μg/l
Simazine	<del>- </del>	0.1 μg/l
2,4-D	<del>- </del>	0.1 μg/l
Total Pesticide		0.5 μg/l
Additional Parameters		
Ammonium	May be naturally present in some water sources and is not harmful.	0.5 mg NH4/l
Ronzono	Benzene may be introduced into source water by industrial effluents or atmospheric pollution. Benzene can	1/1
Benzene	migrate through plastic pipework if petrol is spilled nearby.	1 μg/l
	Benzo(a)pyrene belongs to a group of compounds known as polycyclic aromatic hydrocarbons (PAHs). If	
	detected in drinking water, the usual source is as a result of deterioration of coal tar linings in water mains.	
	Benzo(a)pyrene is seldom detected in drinking water as a result of extensive water mains refurbishment and	
Benzo (a) Pyrene	renewal.	0.01 μg/l
Baran	Very low levels of boron may occur naturally, but in higher amounts could be associated with industrial	1 D/I
Boron	pollution. The standard is health related and has a large safety factor built in.	1 mg B/l
Bromate	Can be associated with industrial pollution or can occur as a by-product of the disinfection process.	10 μg BrO3/l
	Occurs naturally in most water sources. Levels above the standard could give rise to taste issues and contribute	10 μg 5.00/.
Chloride	to corrosion.	250 mg Cl/l
	A measure of the ability of water to conduct an electric current and therefore a measurement of the mineral	2500 μs/cm at
Electrical Conductivity at 20 °C	salts dissolved in the water.	20°C
Nitroto	Nitrate arises from the use of fertilisers from agricultural and may be minimised by good practices and	50 ··· = NO2 /I
Nitrate	appropriate controls. The standard is set well below concentrations that could be harmful.	50 mg NO3/I
	Nitrite may be associated with nitrate or with the use of ammonium in water disinfection. Careful control of the	
	disinfection process reduces formation of nitrite. The standard is set well below concentrations that could be	0.5 mg NO2/I
Nitrite	harmful.	3 ,
	Very low levels of selenium may occur naturally, but in higher amounts could be associated with industrial	
Selenium	pollution. The standard is health related and has a large safety factor built in.	10 μg Se/l
Culphata	Dissolves in water after contact with certain mineral deposits and rock strata. Excess levels can contribute to	250 504/1
Sulphate	corrosion.	250 mg SO4/I
	This standard is the sum of the concentration of trichloroethene and tetrachloroethene. The presence of these	
Sum of Tri & Tetrachloroethene	organic solvents is an indication of industrial pollution.	10 μg/l
		- 1-0/
Tetrachloromethane	The presence of this organic solvent is an indication of industrial pollution.	3 μg/l
		·
	Very low levels of cyanide may occur naturally, but in higher amounts could be associated with industrial	_
Total Cyanide	pollution. The standard is health related and has a large safety factor built in.	50 μg CN/l
Total Organic Carbon	This parameter provides a measure of the total amount of erganic method in water	No abnormal
Total Organic Carbon	This parameter provides a measure of the total amount of organic matter in water.	change
	Associated with the deterioration of old coal tar linings which were used until the mid 1970s. The standards are	
Total PAHs	set well below the levels of significance to health.	0.1 μg/l
	Ĭ	r-O/
	THMs are formed by the reaction of chlorine added as a disinfectant with naturally occurring organic	
	compounds in the water. The standards are set well below the levels of significance to health.	100 μg/l
Total Trihalomethanes	compounds in the water. The standards are set wen below the levels of significance to health.	100 μg/1
Total Trihalomethanes  1, 2 dichloroethane	The presence of this organic solvent is an indication of industrial pollution.	3 μg/l