Water Supply Zone: Staines (AF086) Period: 01-Jan-2024 to 31-Dec-2024

		Nin of		No of Campula-	0/ of Camala-			
Parameter	Units	No. of Samples	PCV	No. of Samples >PCV	% of Samples >PCV	Min.	Mean	Max.
Parameter	Units		Microbiological Parameters	>PCV	>PCV	IVIIII.	IVICALI	IVIAX.
Coliform bacteria	No./100ml	120	0	1	<1	0	0	1
E coli	No./100ml	120	0	1	<1	0	0	1
Clostridium perfringens	No./100ml	36	0	0	0	0	0	0
Enterococci	No./100ml	8	0	0	0	0	0	0
3 day plate count 22 °C	No./1ml at 22 °C	36	No abnormal change	0	0	0	1	7
· ·			Customer Parameters					
Alkalinity	mgHCO3/I	1	No PCV	0	0	215	215	215
Calcium	mgCa/I	1	No PCV	0	0	100	100	100
Chlorine (Residual)	mgCl2/l	120	No PCV	0	0	0.13	0.54	0.97
Colour	mg/I Pt/Co	36	20	0	0	<1.9	<2.5	<2.5
Fluoride	mgF/I	8	1.5	0	0	0.109	0.121	0.135
Hardness (Total)	mgCaCO3/I	1	No PCV	0	0	250	250	250
Hydrogen Ion (pH)	pH value	36	6.5-9.5	0	0	6.9	7.1	7.3
Quantitative Odour	Dilution No.	36	Abnormal & unacceptable to	0	0	0	0	0
Quantitative Taste	Dilution No.	36	consumers	0	0	0	0	0
Temperature	°C	122	No PCV	0	0	6.3	14.5	23.7
Turbidity	NTU	36	4	0	0	<0.10	< 0.10	0.21
			Chemicals					
Metals								
Arsenic	μgAs/l	8	10	0	0	<1.0	<1.0	<1.0
Aluminium	μgAl/l	36	200	0	0	11	22	43.5
Antimony	μgSb/l	8	5	0	0	<0.20	<0.20	0.34
Cadmium	μgCd/l	8	5	0	0	< 0.05	< 0.20	< 0.20
Chromium	μgCr/l	8	50	0	0	< 0.2	<0.5	0.6
Copper	mgCu/l	8	2	0	0	< 0.003	< 0.019	< 0.019
Iron	μgFe/l	36	200	0	0	<3.3	<15.0	26.8
Lead	μgPb/l	8	10	0	0	<0.08	<1.00	2.35
Manganese	μgMn/l	36	50	0	0	< 0.4	<1.0	1.7
Mercury	μgHg/l	8	1	0	0	< 0.02	< 0.10	< 0.10
Nickel	μgNi/l	8	20	0	0	<2.0	<2.0	6.3
Sodium	mgNa/l	8	200	0	0	20.9	27.2	37.1
Pesticides	0 -7							
Atrazine	μg/l	8	0.1	0	0	<0.011	<0.013	<0.013
Carbetamide	μg/l	8	0.1	0	0	< 0.007	< 0.013	< 0.013
Clopyralid	μg/l	8	0.1	0	0	< 0.019	< 0.024	< 0.024
Glyphosate	μg/l	8	0.1	0	0	< 0.005	< 0.005	< 0.005
Mecoprop	μg/l	8	0.1	0	0	< 0.006	< 0.011	< 0.011
Metaldehyde	μg/l	8	0.1	0	0	< 0.009	< 0.021	< 0.021
Metazachlor	μg/l	8	0.1	0	0	<0.008	< 0.011	< 0.011
Propyzamide	μg/l	8	0.1	0	0	< 0.007	< 0.019	< 0.019
Simazine	μg/l	8	0.1	0	0	< 0.009	< 0.017	< 0.017
Total Pesticide	μg/l	8	0.5	0	0	0	0.002	0.012
2,4-D	μg/l	7	0.1	0	0	<0.008	< 0.014	< 0.014
Additional Parameters	P-0/ ·		-		-			
Ammonium	mgNH4/l	8	0.5	0	0	<0.05	<0.05	<0.05
Benzene	μg/l	8	1	0	0	< 0.07	< 0.17	< 0.17
Benzo (a) Pyrene	μg/l	8	0.01	0	0	< 0.001	< 0.001	< 0.001
Boron	mgB/I	8	1	0	0	< 0.045	< 0.100	< 0.100
Bromate	μgBrO3/I	8	10	0	0	<1.5	<1.5	2.2
Chloride	mgCl/l	8	250	0	0	37	46	56
Electrical Conductivity at 20 °C	μS/cm at 20 °C	36	2500	0	0	507	576	637
Nitrate	mgNO3/I	8	50	0	0	16.5	26.8	34.2
Nitrite	mgNO2/l	8	0.5	0	0	<0.007	<0.007	< 0.007
Nitrite Nitrate Formula	0 - 7.	8	1	0	0	<0.33	<0.68	<0.68
Selenium	μgSe/I	8	10	0	0	<0.4	<1.0	<1.0
Sulphate	mgSO4/I	8	250	0	0	64	69	83
Sum of Tri & Tetrachloroethene	μg/l	8	10	0	0	0	0	0
Tetrachloromethane	μg/l	8	3	0	0	<0.2	<0.2	<0.2
Total Cyanide	μgCN/l	9	50	0	0	<6.5	<6.5	<6.5
		8	No abnormal change	0	0	1	2.3	3.3
Total Organic Carbon	mgC/I	9	0.1	0	0	0	0	
Total PAHs	μg/l		100	0	0			0 45.01
Total Trihalomethanes	μg/l	8				15.17	24.4	45.91
1, 2 dichloroethane	μg/l	8	3	0	0	< 0.14	< 0.15	< 0.15

### Notes

PCV = Prescribed Concentration or Value or Specification Concentration or Value

# **Commentary on Water Quality**

In November, a property in Staines detected coliforms and E.coli. Our investigation found that the cause of the failure was due to the condition of the kitchen tap

# **Undertakings & Authorised Departures**

No Authorised Departures applied to this water supply zone during 2024.

# Glossary

The report above show all regulatory parameters which are monitored in accordance with the current Water Supply (Water Quality) Regulations. Some non-regulatory parameters are also included for customer information only. The table below describes these parameters and what the standards can mean.

Regulatory Parameters			
Parameter	What it means	Standard	
Microbiological Parameters			
Coliform bacteria	These bacteria are widely distributed in the environment and provide a sensitive measure of the microbiological quality of the water supply. They are removed during the treatment process. However, if any coliform organisms are detected in drinking water immediate action is taken to investigate the source of the bacteria. Nearly all instances of coliforms in samples taken from customers' taps are due to microbiological growths in the domestic cold taps.	0 per 100ml	
E coli			
Clostridium perfringens Enterococci			
2 day plate count 37 °C	A range of harmless bacteria that may be present in water supplies. These are monitored to ensure the		
3 day plate count 22 °C			
Customer Parameters			
Alkalinity	Alkalinity is normally due to bicarbonate salts of calcium and magnesium, but very occasionally sodium bicarbonate may contribute. In the former case the alkalinity is sometimes called the "temporary hardness" as it is removed by boiling.	No specific standard	
Calcium	Occurs naturally in water after passage through mineral deposits and rock strata. Calcium contributes to the total hardness of water.		
Chlorine (Residual)	Affinity Water disinfects some of our water supplies using chlorine. The concentration of chlorine used is carefully controlled and is set to ensure that water is adequately disinfected, while minimising any taste or odour issues for consumers.	No specific standard	
Colour	Water should be clear and bright, but natural organic matter or pipework corrosion products may occasionally impart a slight tint. The standard is set for reasons of appearance and requires water to be virtually colourless.	20 mg/l Pt/Co	
Fluoride	Occurs naturally in many water sources. The standard is set to ensure no adverse effects. Affinity Water does not artificially fluoridate the water supplies.	1.5 mg F/l	
Hardness (Total)	Hardness is due to calcium and magnesium salts dissolved in the water. Hard water is perfectly safe and there is evidence that it can even be good for your health, unless there are specific requirements to do so there is no need to soften the water. Almost all Affinity Water supplies are hard due to the natural geology of Southern England.	No specific standard	
Hydrogen Ion (pH)	A measure of the acidity or alkalinity of water; pH <7.0 is acidic and pH >7.0 is alkaline. Excessively acidic or alkaline water can contribute to corrosion of pipes and fittings.	Min. 6.5 to max 9.5	
Quantitative Odour Quantitative Taste	Specialist tasting panels examine the water for taste or odour. These standards are measure of the aesthetic quality of drinking water. Unusual odours or tastes may indicate a problem which needs investigating.	Abnormal & unacceptable to consumers	
Temperature		No specific	
Turbidity	The standard requires that there should be no haziness caused by fine particles. Sometimes minute air bubbles give the supply a milky appearance but on standing for a few minutes these will clear from the bottom of the glass upwards.	4 NTU	
Chemicals			
Metals			
Antimony		5 μg Sb/l	
Arsenic		10 μg As/l	
Cadmium	Very low levels of these substances may occur naturally, but in higher amounts could be associated with	5 μg Cd/l	
Chromium	industrial pollution. The standards are health-related and have a large safety factor built in.	50 μg Cr/l	
Mercury		1 μg Hg/l	
Nickel		20 μg Ni/l	
Aluminum	Occurs naturally in many water resources. Aluminum compounds are also used at some water treatment works to remove impurities, but are themselves removed in the process.	200 μg Al/l	
Copper	Any significant amount of copper is likely to come from corrosion of customers' pipes and fittings. An excess of copper can cause a metallic taste.	2 mg Cu/l	
Iron	Iron may be associated with corrosion of old iron water mains. Iron based compounds are also used at some water treatment works to remove impurities, but are themselves removed in the process. The standard for iron has been set for aesthetic reasons as levels persistently above the standard can give rise to discoloured water.	200 μg Fe/l	
Lead	Absent in the water entering supply but variable concentrations of lead may be found in water at the customer's tap in older properties built at a time when lead was commonly used in domestic plumbing systems. The standard recognises that the intake of lead should be minimised for public health reasons.	10 μg Pb/l	

	Occurs naturally in many waters but is usually removed during treatment. The standard is set for aesthetic				
Manganese	reasons as black deposits of manganese dioxide can cause discoloured water.				
	May be naturally present after passing through certain mineral deposits and rock strata or introduced by some				
Sodium	water softening processes. The standard is set well below the level which could affect health.	200 mg Na/l			
Pesticides					
Atrazine		0.1 μg/l			
Carbetamide		0.1 μg/l			
Clopyralid		0.1 μg/l 0.1 μg/l			
Glyphosate Mecoprop	Associated with the use of these substances by agriculture, industry and local authorities. The standards are set				
Metaldehyde	well below the levels that might cause health problems, but levels should be minimised by good practice and				
Metazachlor	appropriate controls. We measure the wide range of substances that may be present.	0.1 μg/l 0.1 μg/l			
Propyzamide		0.1 μg/l			
Simazine		0.1 μg/l			
2,4-D Total Pesticide		0.1 μg/l 0.5 μg/l			
Additional Parameters		0.5 μg/1			
Ammonium	May be naturally present in some water sources and is not harmful.	0.5 mg NH4/l			
	Panzana may be introduced into course water by industrial officents as atmospheric well-time Course				
Benzene	Benzene may be introduced into source water by industrial effluents or atmospheric pollution. Benzene can migrate through plastic pipework if petrol is spilled nearby.	1 μg/l			
Benzene	inigrate through plastic pipework is petron's spineu nearby.	1 μg/1			
	Benzo(a)pyrene belongs to a group of compounds known as polycyclic aromatic hydrocarbons (PAHs). If				
	detected in drinking water, the usual source is as a result of deterioration of coal tar linings in water mains.				
Benzo (a) Pyrene	Benzo(a)pyrene is seldom detected in drinking water as a result of extensive water mains refurbishment and renewal.	0.01 μg/l			
benzo (a) i yiene	i circwai.	0.01 μg/1			
	Very low levels of boron may occur naturally, but in higher amounts could be associated with industrial				
Boron	pollution. The standard is health related and has a large safety factor built in.	1 mg B/l			
Dramata	Control of the district of the	40 003/1			
Bromate	Can be associated with industrial pollution or can occur as a by-product of the disinfection process.	10 μg BrO3/l			
	Occurs naturally in most water sources. Levels above the standard could give rise to taste issues and contribute				
Chloride	to corrosion.	250 mg Cl/l			
Electrical Conductivity at 20 °C	A measure of the ability of water to conduct an electric current and therefore a measurement of the mineral salts dissolved in the water.	2500 μs/cm at 20°C			
Electrical conductivity at 20°C	Saits dissolved in the water.	2500 μ3/cm at 20 C			
	Nitrate arises from the use of fertilisers from agricultural and may be minimised by good practices and				
Nitrate	appropriate controls. The standard is set well below concentrations that could be harmful.	50 mg NO3/I			
	Nikola was kananisha da kata d				
	Nitrite may be associated with nitrate or with the use of ammonium in water disinfection. Careful control of the disinfection process reduces formation of nitrite. The standard is set well below concentrations that could be	0.5 mg NO2/l			
Nitrite	harmful.	0.5 1116 1102/1			
	Very low levels of selenium may occur naturally, but in higher amounts could be associated with industrial				
Selenium	pollution. The standard is health related and has a large safety factor built in.  Dissolves in water after contact with certain mineral deposits and rock strata. Excess levels can contribute to	10 μg Se/l			
Sulphate	corrosion.	250 mg SO4/I			
		0 ,			
	This standard is the sum of the concentration of trichloroethene and tetrachloroethene. The presence of these				
Sum of Tri & Tetrachloroethene	organic solvents is an indication of industrial pollution.	10 μg/l			
Tetrachloromethane	The presence of this organic solvent is an indication of industrial pollution.	3 μg/l			
retractionomethane	The presence of this organic solvent is an indication of industrial pollution.	3 μg/1			
	Very low levels of cyanide may occur naturally, but in higher amounts could be associated with industrial				
Total Cyanide	pollution. The standard is health related and has a large safety factor built in.	50 μg CN/I			
Total Organic Carbon	This parameter provides a measure of the total amount of according to the control of the control	No abnormal			
Total Organic Carbon	This parameter provides a measure of the total amount of organic matter in water.	change			
	Associated with the deterioration of old coal tar linings which were used until the mid 1970s. The standards are				
Total PAHs	set well below the levels of significance to health.	0.1 μg/l			
	TIMe are formed by the reaction of ablasia and date and the fortune to the control of the contro				
Total Trihalomethanes	THMs are formed by the reaction of chlorine added as a disinfectant with naturally occurring organic compounds in the water. The standards are set well below the levels of significance to health.	100 μg/l			
. ca. maiomentales	compounds in the water. The standards are set wen below the levels of significance to health.	100 μg/1			
1, 2 dichloroethane	The presence of this organic solvent is an indication of industrial pollution.	3 μg/l			

Further information can be found on the Affinity Water and Drinking Water Inspectorate websites: https://www.affinitywater.co.uk/index.aspx http://dwi.defra.gov.uk/