Water Supply Zone: Uttlesford Dunmow (AF019) Period: 01-Jan-2024 to 31-Dec-2024

Population: 35810

		No. of	_	No. of Samples	% of Samples			
Parameter	Units	Samples	PCV	>PCV	>PCV	Min.	Mean	Max.
			Aicrobiological Parameters					
Coliform bacteria	No./100ml	96	0	0	0	0	0	0
E coli	No./100ml	96	0	0	0	0	0	0
Clostridium perfringens	No./100ml	36	0	0	0	0	0	0
Enterococci	No./100ml	8	0	0	0	0	0	0
3 day plate count 22 °C	No./1ml at 22 °C	36	No abnormal change	0	0	0	1	29
A11 11 11	11000/1		Customer Parameters	0	0	252	252	252
Alkalinity	mgHCO3/I	1	No PCV	0		353	353	353
Calcium	mgCa/l	1	No PCV No PCV	0 0	0 0	137	137	137
Chlorine (Residual)	mgCl2/l	96				0.03	0.25	0.61
Colour	mg/l Pt/Co	36	20	0	0	<1.9	<2.5	<2.5
Fluoride	mgF/l	8	1.5	0	0	0.187	0.513	1.251
Hardness (Total)	mgCaCO3/I	1	No PCV	0	0	343	343	343
Hydrogen Ion (pH)	pH value	36	6.5-9.5	0	0	6.9	7.1	8
Quantitative Odour	Dilution No.	36	Abnormal & unacceptable to	1	3	0	0	1
Quantitative Taste	Dilution No.	36	consumers	0	0	0	0	0
Temperature	°C	97	No PCV	0	0	7.6	13.9	22.6
Turbidity	NTU	36	4	0	0	<0.10	<0.10	0.21
			Chemicals					
Metals								
Arsenic	µgAs/l	8	10	0	0	<0.1	<1.0	1.5
Aluminium	μgAl/l	36	200	0	0	<2.0	<5.0	<5.0
Antimony	µgSb/l	8	5	0	0	<0.03	<0.20	<0.20
Cadmium	μgCd/l	8	5	0	0	<0.05	<0.20	<0.20
Chromium	μgCr/l	8	50	0	0	<0.2	0.2	0.8
Copper	mgCu/l	8	2	0	0	0.004	0.029	0.068
Iron	μgFe/l	36	200	0	0	<3.3	<15.0	<15.0
Lead	µgPb/l	8	10	0	0	<1.00	<1.00	1.38
Manganese	μgMn/l	36	50	0	0	<0.4	<1.0	<1.0
Mercury	μgHg/l	8	1	0	0	<0.02	<0.10	<0.10
Nickel	μgNi/l	8	20	0	0	0.4	1.5	2.3
Sodium	mgNa/l	8	200	0	0	11.7	31.8	82.8
Pesticides	0 .,							
Atrazine	μg/l	8	0.1	0	0	<0.011	< 0.013	< 0.013
Total Pesticide	μg/l	8	0.5	0	0	0	0	0
Additional Parameters	10,							
Ammonium	mgNH4/I	36	0.5	0	0	<0.05	< 0.05	< 0.05
Benzene	μg/l	9	1	0	0	<0.07	<0.17	<0.17
Benzo (a) Pyrene	μg/l	8	0.01	0	0	<0.001	< 0.001	<0.001
Boron	mgB/I	8	1	0	0	<0.045	<0.100	0.2
Bromate	μgBrO3/I	8	10	0	0	<1.5	<1.5	1.5
Chloride	mgCl/l	8	250	0	0	24	34	61
Electrical Conductivity at 20 °C	μS/cm at 20 °C	36	2500	0	0	542	614	791
Nitrate	mgNO3/I	36	50	0	0	<3.4	14.8	20.4
Nitrite	mgNO2/I	36	0.5	0	0	<0.007	<0.007	<0.007
Nitrite Nitrate Formula	mgin02/1	8	1	0	0	<0.007	<0.38	<0.38
Selenium	ugSo/l	8	10	0	0	<0.00	<0.38	<1.0
	μgSe/l			0	0			<1.0 60
Sulphate	mgSO4/l	8	250			24	39	
Sum of Tri & Tetrachloroethene	μg/l	9	10	0	0	0	0	0
Tetrachloromethane	μg/l	9	3	0	0	<0.2	<0.2	<0.2
Total Cyanide	μgCN/I	8	50	0	0	<6.5	<6.5	<6.5
Total Organic Carbon	mgC/I	8	No abnormal change	0	0	0.4	0.5	0.7
Total PAHs	μg/l	8	0.1	0	0	0	0	0
Total Trihalomethanes	μg/l	9	100	0	0	2.77	6.14	8.69
1, 2 dichloroethane	μg/l	8	3	0	0	<0.14	<0.15	<0.15

Notes

PCV = Prescribed Concentration or Value or Specification Concentration or Value

Commentary on Water Quality

In August, a solvent odour was detected in a property in Dunmow. Our investigation found the cause to be low turnover of water. We have reconfigured the network and carried out remedial works to reduce a re-occurrence

Undertakings & Authorised Departures

No Undertakings or Authorised Departures applied to this water supply zone during 2024

Glossary Drinking Water Standards

The report above show all regulatory parameters which are monitored in accordance with the current Water Supply (Water Quality) Regulations. Some non-regulatory parameters are also included for customer information only. The table below describes these parameters and what the standards can mean.

Regulatory Parameters

Regulatory Parameters		
Parameter	What it means	Standard
Microbiological Parameters		
Coliform bacteria	These bacteria are widely distributed in the environment and provide a sensitive measure of the microbiological quality of the water supply. They are removed during the treatment process. However, if any coliform organisms are detected in drinking water immediate action is taken to investigate the source of the bacteria. Nearly all instances of coliforms in samples taken from customers' taps are due to microbiological growths in the domestic cold taps.	
E coli		0 per 100ml
Clostridium perfringens		
Enterococci	Bacteria which are indicative of possible faecal contamination. Immediate action is taken if these organisms are	
	detected in drinking water.	0 per 100ml
		No specific
		standard
2 day plate count 37 °C	A range of harmless bacteria that may be present in water supplies. These are monitored to ensure the	(increasing trends
3 day plate count 22 °C	efficiency of the treatment process and the cleanliness of the distribution system.	are investigated)
Customer Parameters		
	Alkalinity is normally due to bicarbonate salts of calcium and magnesium, but very occasionally sodium	
	bicarbonate may contribute. In the former case the alkalinity is sometimes called the "temporary hardness" as it	No specific
Alkalinity	is removed by boiling.	standard
	Occurs naturally in water after passage through mineral deposits and rock strata. Calcium contributes to the	No specific
Calcium	total hardness of water.	standard
	Affinity Water disinfects some of our water supplies using chlorine. The concentration of chlorine used is	
	carefully controlled and is set to ensure that water is adequately disinfected, while minimising any taste or	No specific
Chlorine (Residual)	odour issues for consumers.	standard
	Water should be clear and bright, but natural organic matter or pipework corrosion products may occasionally	
Colour	impart a slight tint. The standard is set for reasons of appearance and requires water to be virtually colourless.	20 mg/l Pt/Co
	Occurs naturally in many water sources. The standard is set to ensure no adverse effects. Affinity Water does	
Fluoride	not artificially fluoridate the water supplies.	1.5 mg F/l
	Hardness is due to calcium and magnesium salts dissolved in the water. Hard water is perfectly safe and there is	
	evidence that it can even be good for your health, unless there are specific requirements to do so there is no	N
Llardnass (Tatal)	need to soften the water. Almost all Affinity Water supplies are hard due to the natural geology of Southern	No specific
Hardness (Total)	England.	standard
	A measure of the acidity or alkalinity of water; pH <7.0 is acidic and pH >7.0 is alkaline. Excessively acidic or	
Hydrogen Ion (pH)	alkaline water can contribute to corrosion of pipes and fittings.	Min. 6.5 to max 9.5
inverogen ion (pri)		Abnormal &
Quantitative Odour	Specialist tasting panels examine the water for taste or odour. These standards are measure of the aesthetic	unacceptable to
Quantitative Taste	quality of drinking water. Unusual odours or tastes may indicate a problem which needs investigating.	consumers
Temperature		No specific
	The standard requires that there should be no haziness caused by fine particles. Sometimes minute air bubbles	no specifie
	give the supply a milky appearance but on standing for a few minutes these will clear from the bottom of the	4 NTU
Turbidity	glass upwards.	
Chemicals		
Metals		
Antimony		5 μg Sb/l
Arsenic]	10 µg As/l
Cadmium	Very low levels of these substances may occur naturally, but in higher amounts could be associated with	5 μg Cd/l
Chromium	industrial pollution. The standards are health-related and have a large safety factor built in.	50 μg Cr/l
Mercury		1 μg Hg/l
Nickel		20 μg Ni/l
	Occurs naturally in many water resources. Aluminum compounds are also used at some water treatment works	202
Aluminum	to remove impurities, but are themselves removed in the process.	200 µg Al/l
	Any configuration and the conner is likely to some from connection of evotors and states and finite or the	
Coppor	Any significant amount of copper is likely to come from corrosion of customers' pipes and fittings. An excess of	2 mg C /
Copper	copper can cause a metallic taste.	2 mg Cu/l
	Iron may be accorded with correction of old iron water mains. Iron based compounds are also used at some	
	Iron may be associated with corrosion of old iron water mains. Iron based compounds are also used at some water treatment works to remove impurities, but are themselves removed in the process. The standard for iron	
Iron	has been set for aesthetic reasons as levels persistently above the standard can give rise to discoloured water.	200 µg Fe/l
	nas seen see on sestiene reasons as revers persistently above, the standard can give rise to discoloured water.	200 µ5 1 C/ 1
	Absent in the water entering supply but variable concentrations of lead may be found in water at the customer's	
	tap in older properties built at a time when lead was commonly used in domestic plumbing systems. The	
Lead	standard recognises that the intake of lead should be minimised for public health reasons.	10 µg Pb/l
	Occurs naturally in many waters but is usually removed during treatment. The standard is set for aesthetic	
Manganese	reasons as black deposits of manganese dioxide can cause discoloured water.	50µg Mn/l
	May be naturally present after passing through certain mineral deposits and rock strata or introduced by some	
Sodium	water softening processes. The standard is set well below the level which could affect health.	200 mg Na/l

Pesticides					
Atrazine		0.1 μg/l			
Carbetamide					
		0.1 μg/l			
Clopyralid		0.1 μg/l			
Glyphosate		0.1 μg/l 0.1 μg/l			
Mecoprop	hyde well below the levels that might cause health problems, but levels should be minimised by good practice and				
Metaldehyde					
Metazachlor					
Propyzamide		0.1 μg/l			
Simazine	—	0.1 μg/l			
2,4-D	—	0.1 μg/l			
Total Pesticide	—	0.5 μg/l			
Additional Parameters		0.5 μg/1			
Ammonium	May be naturally present in some water sources and is not harmful.	0.5 mg NH4/l			
Benzene	Benzene may be introduced into source water by industrial effluents or atmospheric pollution. Benzene can migrate through plastic pipework if petrol is spilled nearby.	1 µg/l			
Benzo (a) Pyrene	Benzo(a)pyrene belongs to a group of compounds known as polycyclic aromatic hydrocarbons (PAHs). If detected in drinking water, the usual source is as a result of deterioration of coal tar linings in water mains. Benzo(a)pyrene is seldom detected in drinking water as a result of extensive water mains refurbishment and renewal.	0.01 µg/l			
Boron	Very low levels of boron may occur naturally, but in higher amounts could be associated with industrial pollution. The standard is health related and has a large safety factor built in.	1 mg B/l			
Bromate	Can be associated with industrial pollution or can occur as a by-product of the disinfection process.	10 µg BrO3/I			
Chloride	Occurs naturally in most water sources. Levels above the standard could give rise to taste issues and contribute to corrosion.	250 mg Cl/l			
Electrical Conductivity at 20 °C	A measure of the ability of water to conduct an electric current and therefore a measurement of the mineral salts dissolved in the water.	2500 μs/cm at 20°C			
Nitrate	Nitrate arises from the use of fertilisers from agricultural and may be minimised by good practices and appropriate controls. The standard is set well below concentrations that could be harmful.	50 mg NO3/I			
Nitrite	Nitrite may be associated with nitrate or with the use of ammonium in water disinfection. Careful control of the disinfection process reduces formation of nitrite. The standard is set well below concentrations that could be harmful.	0.5 mg NO2/I			
Selenium	Very low levels of selenium may occur naturally, but in higher amounts could be associated with industrial pollution. The standard is health related and has a large safety factor built in.	10 µg Se/l			
Sulphate	Dissolves in water after contact with certain mineral deposits and rock strata. Excess levels can contribute to corrosion.	250 mg SO4/I			
Sum of Tri & Tetrachloroethene	This standard is the sum of the concentration of trichloroethene and tetrachloroethene. The presence of these organic solvents is an indication of industrial pollution.	10 μg/l			
Tetrachloromethane	The presence of this organic solvent is an indication of industrial pollution.	3 µg/l			
Total Cyanide	Very low levels of cyanide may occur naturally, but in higher amounts could be associated with industrial pollution. The standard is health related and has a large safety factor built in.	50 μg CN/I			
Total Organic Carbon	This parameter provides a measure of the total amount of organic matter in water.	No abnormal change			
Total PAHs	Associated with the deterioration of old coal tar linings which were used until the mid 1970s. The standards are set well below the levels of significance to health.	0.1 μg/l			
Total Trihalomethanes	THMs are formed by the reaction of chlorine added as a disinfectant with naturally occurring organic compounds in the water. The standards are set well below the levels of significance to health.	100 µg/l			
1, 2 dichloroethane	The presence of this organic solvent is an indication of industrial pollution.	3 μg/l			

Further information can be found on the Affinity Water and Drinking Water Inspectorate websites: https://www.affinitywater.co.uk/index.aspx

http://dwi.defra.gov.uk/