Water Supply Zone: Royston (AF003) Period: 01-Jan-2024 to 31-Dec-2024

Population: 17142

		No. of		No. of Samples	% of Samples			
Parameter	Units	Samples	PCV	>PCV	>PCV	Min.	Mean	Max.
rarameter	Ullits		licrobiological Parameters	>FCV	>FCV	IVIIII.	IVICALI	IVIAA.
Coliform bacteria	No./100ml	48	0	0	0	0	0	0
E coli	No./100ml	48	0	0	0	0	0	0
Clostridium perfringens	No./100ml	8	0	0	0	0	0	0
Enterococci	No./100ml	8	0	0	0	0	0	0
3 day plate count 22 °C	No./1ml at 22 °C	24	No abnormal change	0	0	0	0	1
3 day plate coulit 22 C	NO./IIII at 22 C	24	Customer Parameters					
Alkalinity	mgHCO3/I	1	No PCV	0	0	294	294	294
Calcium	mgCa/I	1	No PCV	0	0	118	118	118
Chlorine (Residual)	mgCl2/l	48	No PCV	0	0	0.05	0.3	0.56
Colour	mg/l Pt/Co	24	20	0	0	<1.9	<2.5	<2.5
Fluoride	mgF/I	8	1.5	0	0	0.128	0.149	0.16
Hardness (Total)	mgCaCO3/I	1	No PCV	0	0	295	295	295
Hydrogen Ion (pH)	-	24	6.5-9.5	0	0	293 7	7.2	7.3
	pH value	24		0	0	0	0	7.3 0
Quantitative Odour	Dilution No.		Abnormal & unacceptable to	-	~			
Quantitative Taste	Dilution No.	24	consumers No PCV	0	0 0	0	0	0
Temperature	°C	48				8	13.4	19.3
Turbidity	NTU	24	4 Chamilanta	0	0	<0.10	<0.10	0.15
Metals			Chemicals					
Arsenic	μgAs/l	8	10	0	0	<1.0	<1.0	<1.0
Aluminium	μgAI/I	8	200	0	0	<2.0	<5.0	<5.0
Antimony		8	5	0	0	<0.20	<0.20	<0.20
Cadmium	μgSb/l	8	5	0	0	<0.20	<0.20	<0.20
	μgCd/l	8	50	0	0	<0.05	<0.20	
Chromium	μgCr/l							1.1
Copper	mgCu/l	8	2	0	0	<0.019	0.046	0.23
Iron	μgFe/l	8	200	0	0	<3.3	<15.0	<15.0
Lead	μgPb/l	8	10	0	0	<1.00	<1.00	1.98
Manganese	μgMn/l	8	50	0	0	<0.4	<1.0	<1.0
Mercury	μgHg/l	8	1	0	0	<0.02	<0.10	<0.10
Nickel	μgNi/l	8	20	0	0	1.1	2.2	3
Sodium	mgNa/l	8	200	0	0	10	11.3	14.2
Pesticides						0.011	0.010	
Atrazine	μg/l	8	0.1	0	0	<0.011	<0.013	<0.013
Total Pesticide	μg/l	8	0.5	0	0	0	0.01	0.012
Additional Parameters Ammonium	mgNH4/l	8	0.5	0	0	<0.05	<0.05	<0.05
Benzene	μg/l	8	1	0	0	<0.03	<0.03	<0.03
		8	0.01	0	0	<0.001	<0.17	<0.17
Benzo (a) Pyrene	μg/l	8		0	0			
Boron	mgB/I		1	0	0	<0.045	<0.100	<0.100
Bromate	μgBrO3/l	8	10			<1.5	<1.5	<1.5
Chloride	mgCl/l	8	250	0	0	22	27	44
Electrical Conductivity at 20 °C	μS/cm at 20 °C	24	2500	0	0	490	520	582
Nitrate	mgNO3/I	8	50	0	0	30.4	34.3	38.9
Nitrite	mgNO2/I	8	0.5	0	0	<0.007	<0.007	<0.007
Nitrite Nitrate Formula		8	1	0	0	<0.61	<0.78	<0.78
Selenium	μgSe/l	8	10	0	0	<1.0	<1.0	<1.0
Sulphate	mgSO4/I	8	250	0	0	18	23	35
Sum of Tri & Tetrachloroethene	μg/l	8	10	0	0	0	0.6	0.8
Tetrachloromethane	μg/l	8	3	0	0	<0.2	<0.2	< 0.2
Total Cyanide	μgCN/I	8	50	0	0	<6.5	<6.5	<6.5
Total Organic Carbon	mgC/I	8	No abnormal change	0	0	0.5	0.6	0.9
Total PAHs	μg/l	9	0.1	0	0	0	0	0
Total Trihalomethanes	μg/l	8	100	0	0	3.08	5.59	9.2
1, 2 dichloroethane	μg/l	8	3	0	0	< 0.14	<0.15	< 0.15

# Notes

PCV = Prescribed Concentration or Value or Specification Concentration or Value

## **Commentary on Water Quality**

Water quality was satisfactory in this zone in 2024.

## **Undertakings & Authorised Departures**

No Undertakings or Authorised Departures applied to this water supply zone during 2024  $\,$ 

Glossary

Drinking Water Standards

The report above show all regulatory parameters which are monitored in accordance with the current Water Supply (Water Quality) Regulations. Some non-regulatory parameters are also included for customer information only. The table below describes these parameters and what the standards can mean.

Regulatory Parameters		
Parameter	What it means	Standard
Microbiological Parameters		
	These bacteria are widely distributed in the environment and provide a sensitive measure of the microbiological quality of the water supply. They are removed during the treatment process. However, if any coliform organisms are detected in drinking water immediate action is taken to investigate the source of the bacteria. Nearly all instances of coliforms in samples taken from customers' taps are due to microbiological growths in	
Coliform bacteria	the domestic cold taps.	0 per 100ml
E coli Clostridium perfringens Enterococci	Bacteria which are indicative of possible faecal contamination. Immediate action is taken if these organisms are detected in drinking water.	0 per 100ml
2 day plate count 37 °C 3 day plate count 22 °C Customer Parameters	A range of harmless bacteria that may be present in water supplies. These are monitored to ensure the efficiency of the treatment process and the cleanliness of the distribution system.	No specific standard (increasing trends are investigated)
customer Parameters	All of the transport of the decay of the dec	
Alkalinity	Alkalinity is normally due to bicarbonate salts of calcium and magnesium, but very occasionally sodium bicarbonate may contribute. In the former case the alkalinity is sometimes called the "temporary hardness" as it is removed by boiling.	No specific standard
Calcium	Occurs naturally in water after passage through mineral deposits and rock strata. Calcium contributes to the total hardness of water.	No specific standard
Chlorine (Residual)	Affinity Water disinfects some of our water supplies using chlorine. The concentration of chlorine used is carefully controlled and is set to ensure that water is adequately disinfected, while minimising any taste or odour issues for consumers.	No specific standard
Colour	Water should be clear and bright, but natural organic matter or pipework corrosion products may occasionally impart a slight tint. The standard is set for reasons of appearance and requires water to be virtually colourless.	20 mg/l Pt/Co
Fluoride	Occurs naturally in many water sources. The standard is set to ensure no adverse effects. Affinity Water does not artificially fluoridate the water supplies.	1.5 mg F/l
Hardness (Total)	Hardness is due to calcium and magnesium salts dissolved in the water. Hard water is perfectly safe and there is evidence that it can even be good for your health, unless there are specific requirements to do so there is no need to soften the water. Almost all Affinity Water supplies are hard due to the natural geology of Southern England.	No specific standard
Hydrogen Ion (pH)	A measure of the acidity or alkalinity of water; pH <7.0 is acidic and pH >7.0 is alkaline. Excessively acidic or alkaline water can contribute to corrosion of pipes and fittings.	Min. 6.5 to max 9.5
Quantitative Odour Quantitative Taste Temperature	Specialist tasting panels examine the water for taste or odour. These standards are measure of the aesthetic quality of drinking water. Unusual odours or tastes may indicate a problem which needs investigating.	Abnormal & unacceptable to consumers
Turbidity	The standard requires that there should be no haziness caused by fine particles. Sometimes minute air bubbles give the supply a milky appearance but on standing for a few minutes these will clear from the bottom of the glass upwards.	No specific 4 NTU
Chemicals	John Str. Communication (Communication Communication Commu	
Metals		
Antimony Arsenic Cadmium Chromium Mercury Nickel	Very low levels of these substances may occur naturally, but in higher amounts could be associated with industrial pollution. The standards are health-related and have a large safety factor built in.	5 µg Sb/l 10 µg As/l 5 µg Cd/l 50 µg Cr/l 1 µg Hg/l 20 µg Ni/l
Aluminum	Occurs naturally in many water resources. Aluminum compounds are also used at some water treatment works to remove impurities, but are themselves removed in the process.	200 μg Al/l
Copper	Any significant amount of copper is likely to come from corrosion of customers' pipes and fittings. An excess of copper can cause a metallic taste.	2 mg Cu/l
Iron	Iron may be associated with corrosion of old iron water mains. Iron based compounds are also used at some water treatment works to remove impurities, but are themselves removed in the process. The standard for iron has been set for aesthetic reasons as levels persistently above the standard can give rise to discoloured water.	200 μg Fe/l
Lead	Absent in the water entering supply but variable concentrations of lead may be found in water at the customer's tap in older properties built at a time when lead was commonly used in domestic plumbing systems. The standard recognises that the intake of lead should be minimised for public health reasons.	10 μg Pb/l
Manganese	Occurs naturally in many waters but is usually removed during treatment. The standard is set for aesthetic reasons as black deposits of manganese dioxide can cause discoloured water.	50μg Mn/l
Sodium	May be naturally present after passing through certain mineral deposits and rock strata or introduced by some water softening processes. The standard is set well below the level which could affect health.	200 mg Na/l

Pesticides		
Atrazine		0.1 μg/l
Carbetamide	-	
	-	0.1 μg/l 0.1 μg/l
Clumbasata	-	
Glyphosate		0.1 μg/l
Mecoprop	Associated with the use of these substances by agriculture, industry and local authorities. The standards are set	0.1 μg/l
Metaldehyde	well below the levels that might cause health problems, but levels should be minimised by good practice and	0.1 μg/l
Metazachlor	appropriate controls. We measure the wide range of substances that may be present.	0.1 μg/l
Propyzamide		0.1 μg/l
Simazine	<u></u>	0.1 μg/l
2,4-D		0.1 μg/l
Total Pesticide		0.5 μg/l
Additional Parameters		
l		
Ammonium	May be naturally present in some water sources and is not harmful.	0.5 mg NH4/I
Benzene	Benzene may be introduced into source water by industrial effluents or atmospheric pollution. Benzene can migrate through plastic pipework if petrol is spilled nearby.	1 μg/l
Benzo (a) Pyrene	Benzo(a)pyrene belongs to a group of compounds known as polycyclic aromatic hydrocarbons (PAHs). If detected in drinking water, the usual source is as a result of deterioration of coal tar linings in water mains. Benzo(a)pyrene is seldom detected in drinking water as a result of extensive water mains refurbishment and renewal.	0.01 μg/l
	Very low levels of boron may occur naturally, but in higher amounts could be associated with industrial	
Boron	pollution. The standard is health related and has a large safety factor built in.	1 mg B/l
Bromate	Can be associated with industrial pollution or can occur as a by-product of the disinfection process.	10 μg BrO3/l
Chloride	Occurs naturally in most water sources. Levels above the standard could give rise to taste issues and contribute to corrosion.	250 mg Cl/l
	to controsion	230 1116 0171
Electrical Conductivity at 20 °C	A measure of the ability of water to conduct an electric current and therefore a measurement of the mineral salts dissolved in the water.	2500 μs/cm at 20°C
Nitrate	Nitrate arises from the use of fertilisers from agricultural and may be minimised by good practices and appropriate controls. The standard is set well below concentrations that could be harmful.	50 mg NO3/l
Nitrite	Nitrite may be associated with nitrate or with the use of ammonium in water disinfection. Careful control of the disinfection process reduces formation of nitrite. The standard is set well below concentrations that could be harmful.	0.5 mg NO2/l
	Very low levels of selenium may occur naturally, but in higher amounts could be associated with industrial	
Selenium	pollution. The standard is health related and has a large safety factor built in.	10 μg Se/l
Sulphate	Dissolves in water after contact with certain mineral deposits and rock strata. Excess levels can contribute to corrosion.	250 mg SO4/I
Sum of Tri & Tetrachloroethene	This standard is the sum of the concentration of trichloroethene and tetrachloroethene. The presence of these organic solvents is an indication of industrial pollution.	10 μg/l
Tetrachloromethane	The presence of this organic solvent is an indication of industrial pollution.	3 μg/l
Total Cyanide	Very low levels of cyanide may occur naturally, but in higher amounts could be associated with industrial pollution. The standard is health related and has a large safety factor built in.	50 μg CN/I
Total Organic Carbon	This parameter provides a measure of the total amount of organic matter in water.	No abnormal change
Total PAHs	Associated with the deterioration of old coal tar linings which were used until the mid 1970s. The standards are set well below the levels of significance to health.	0.1 μg/l
Total Trihalomethanes	THMs are formed by the reaction of chlorine added as a disinfectant with naturally occurring organic compounds in the water. The standards are set well below the levels of significance to health.	100 μg/l
1, 2 dichloroethane	The presence of this organic solvent is an indication of industrial pollution.	3 μg/l

Further information can be found on the Affinity Water and Drinking Water Inspectorate websites: https://www.affinitywater.co.uk/index.aspx http://dwi.defra.gov.uk/